



ZIAUDDIN UNIVERSITY
EXAMINATION BOARD

Higher Secondary School Certificate (HSSC)

Examination Syllabus

Elementary Chemistry and Chemical Pathology XII

**Based on Curriculum of Medical
Technology developed by
Curriculum Wing Ministry of Education,
Islamabad**

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PREFACE

The Ziauddin University Examination Board (ZUEB) was established under **Sindh ACT XLI 2018**, with the primary objective of enhancing the quality of education in Sindh. ZUEB is responsible for administering examinations for the **Secondary School Certificate (SSC)** and **Higher Secondary School Certificate (HSSC)** in alignment with the curriculum of Medical Technology set by the **Curriculum Wing, Ministry of Education, Islamabad**. Through its ordinance, ZUEB is mandated to provide examination services for both English, Urdu, and Sindhi medium candidates from private schools across Sindh. This examination syllabus reflects ZUEB's dedication to achieving the educational goals set by the provincial authorities.

In collaboration with subject professors, ZUEB has developed a comprehensive syllabus for each subject. It is important to distinguish between the syllabus and the curriculum. The syllabus serves as a guide for both teachers and students, outlining the key areas of focus within the subject. It provides students with a clear understanding of what is expected of them in their studies and helps them prepare effectively for their exams.

This examination syllabus incorporates all cognitive outcomes derived from the **Curriculum of Medical Technology**, ensuring that assessments are both valid and reliable. While the focus is primarily on the cognitive domain, significant emphasis is placed on the application of knowledge and understanding.

The syllabus is made available to all stakeholders via the ZUEB website to assist affiliated schools in planning their teaching. It is crucial to note that the syllabus, rather than the prescribed textbook, forms the foundation of ZUEB examinations. Additionally, this syllabus supports the development of learning materials for both students and teachers. ZUEB remains committed to supporting students undertaking the SSC and HSSC courses by facilitating their learning outcomes through this detailed syllabus document.

To further assist in the learning process, ZUEB plans to provide a dedicated **e-resource tab** on its website, offering both text-based and video content on various subjects. These 15-20 minute instructional videos, created around key subject concepts, allow students to learn at their own pace and convenience. The videos can be used as a reinforcement tool to revisit lessons already taught or as pre-lesson material. This initiative is an ongoing effort, and new videos will continue to be uploaded.

We encourage all students and educators to make the most of these resources for a more enriched and flexible learning experience.

Sincerely,
Shahbaz Nasim
Head – Measurement & Testing
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Reviewed by:
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Rationale For The Reviewed Provincial Curriculum

The process of revising the National Curriculum 2006 began in August 2004, when the newly elected government of Pakistan initiated education reforms across the country. These reforms included the introduction of a new National Education Policy, a National Education Census, and a revision of curricula (Ministry of Education, 2009).

In practice, the overhaul of the secondary school curriculum began in 2006, leading to a review of the scheme of studies for classes I to XII and the revision of curricula for 25 compulsory subjects.

The 18th Amendment to the Constitution of Pakistan, enacted in 2010, significantly altered the federal-provincial relationship by abolishing the "concurrent legislative list." This amendment granted provinces greater legislative and financial autonomy in sectors such as education and health. The most notable implication of the 18th Amendment for education was the transfer of responsibility for curriculum development, syllabus planning, policy formation, and educational standards to the provinces, marking a significant step forward for education.

The School Education Department tasked a Curriculum Review Team with revising the National Curriculum 2006 for all subjects. The goal was to create a curriculum better suited to the needs of students and teachers while aligning with the principles of the 18th Amendment. Subject-specific curriculum review committees were established to critically examine and align the curriculum's content, both contextually and textually, ensuring coherence across various subjects. The Bureau of Curriculum (BoC) played a crucial role in organizing workshops and meetings in Hyderabad to facilitate the completion of this task. The support of numerous educationists, researchers, and teachers was invaluable in successfully revising the curriculum.

The Ziauddin University Examination Board (ZUEB) SSC and HSSC syllabi are developed in accordance with the Revised Curriculum. To date, textbooks for various subjects have been developed based on the revised curriculum.

ZIAUDDIN UNIVERSITY EXAMINATION BOARD
SLOs CATEGORIZATION
XII-Elementary Chemistry and Chemical Pathology
Detailed Syllabus

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
A	Introduction to Chemistry	A-01	Define chemistry and explain its basic concepts.	✓		
		A-02	Identify and list the main divisions (branches) of chemistry.	✓		
		A-03	Differentiate between various branches of chemistry based on their area of study.		✓	
		A-04	Explain the practical utility and applications of each branch of chemistry in daily life and industries.			✓
		A-05	Describe the relationship of chemistry with other sciences such as biology, physics, and environmental science.		✓	
		A-06	Identify real-life examples of the use of chemistry in health, agriculture, food, and environment.			✓
B	Elements	B-01	Define element, atom, molecule, and compound.	✓		
		B-02	Describe the structure of an atom with subatomic particles.		✓	
		B-03	Explain the arrangement of elements in the periodic table and their classification.		✓	

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
		B-04	Apply the concept of valency to predict bonding behavior of elements.			✓
		B-05	Identify symbols of important elements and their valencies.	✓		
		B-06	Differentiate between metals, nonmetals, and metalloids based on their properties.		✓	
		B-07	Analyze periodic trends to classify elements into s, p, d, and f blocks.			✓
		B-08	Relate the uses of important elements to their properties.			✓
C	Compounds and Mixtures	C-01	Define compounds and mixtures with examples.	✓	✓	
		C-02	Distinguish between compound and mixture based on their properties.			✓
		C-03	Describe methods used for the separation of mixtures.		✓	
		C-04	Classify compounds into molecular and ionic types with examples.		✓	
		C-05	Explain the importance of chemical composition in identifying pure substances.		✓	
		C-06	Describe and differentiate types of distillation.		✓	
		C-07	Explain Raoult's Law and Dalton's Law and their role in distillation.			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
		C-08	Describe filtration and its types with examples.		✓	
		C-09	Explain sedimentation and decantation as physical separation methods.		✓	
		C-10	List common chemical compounds along with their chemical formulas.	✓		
D	Units of Measurements	D-01	Define measurement and explain its importance in chemistry and science.		✓	
		D-02	Distinguish between metric and imperial systems of measurement.		✓	
		D-03	Convert units between Celsius and Fahrenheit temperature scales.			✓
		D-04	Define physical quantities and differentiate between fundamental and derived physical quantities.		✓	
		D-05	Identify SI units and measuring instruments for length, mass, volume, temperature, energy, and heat.			✓
		D-06	Perform simple calculations using derived quantities such as speed, density, pressure, and energy.			✓
		D-07	Use appropriate instruments for measurement in basic chemistry experiments.			✓
		D-08	Measurement of weight in different units (Practical).			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
		D-09	Measurement of volume in different units (Practical).			✓
		D-10	Measurement of length in different units (Practical).			✓
		D-11	Measurement of heat in different units (Practical).			✓
E	Solution	E-01	Define solution and explain its properties with appropriate examples.		✓	
		E-02	Differentiate between compounds and solutions based on composition and separation.		✓	
		E-03	Classify solutions by solute/solvent state and describe types like saturated, unsaturated, and supersaturated.		✓	
		E-04	Explain isotonic, hypertonic, and hypotonic solutions and their effects on cells.		✓	
		E-05	Define solubility and describe the key factors influencing it (temperature, pressure, bonding).		✓	
		E-06	Define and compare different concentration terms: molarity, molality, normality, and formality.		✓	
		E-07	Perform step-by-step preparation of a standard solution with accurate measurements.			✓
		E-08	Solubility - demonstration of factors affecting solubility (Practical).			✓
		E-09	Preparation of concentrated			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			solution (Practical).			
		E-10	Preparation of Molar solution of different compounds (Practical).			✓
		E-11	Preparation of saturated solution of different compounds (Practical).			✓
F	Acid, Base and Salt	F-01	Define acids, bases, salts, and alkalis and describe their properties.		✓	
		F-02	Describe the reactions of acids with metals, metal carbonates, and bicarbonates.		✓	
		F-03	Classify acids and bases based on concentration, ionization, and acidity.		✓	
		F-04	Differentiate between Arrhenius, Bronsted-Lowry, and Lewis concepts of acids and bases.			✓
		F-05	Describe physical and chemical properties of salts and classify types of salts.		✓	
		F-06	Explain the concept of pH and the strength of acids and bases using the pH scale.			✓
		F-07	Identify and compare natural, synthetic, and olfactory indicators.		✓	
		F-08	Explain the process of titration and determine equivalence point using indicators.			✓
		F-09	Apply the Henderson-Hasselbalch equation to calculate pH of buffer			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			solutions.			
G	Various Chemical Processes	G-01	Define various chemical and physical processes in chemistry.	✓		
		G-02	Describe the procedure of key chemical processes such as filtration, distillation, etc.		✓	
		G-03	Classify different types of chemical separation methods.		✓	
		G-04	Explain the utilization of each process in real-life applications.			✓
		G-05	Analyze the importance of catalysts and redox reactions in chemical industries.			✓
		G-06	Perform basic chemical operations such as titration and filtration in laboratory scenarios.			✓
H	Chemical Reactions	H-01	Define chemical reactions and describe their characteristics.		✓	
		H-02	Identify the reactants and products in a chemical equation.		✓	
		H-03	Classify chemical reactions into their respective types.			✓
		H-04	Differentiate between combination, decomposition, displacement, and double displacement reactions.			✓
		H-05	Explain real-life applications of various types of chemical reactions.			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
		H-06	Balance basic chemical equations related to different types of reactions.			✓
I	Introduction to Chemical Pathology	I-01	Define chemical pathology and describe its role in disease diagnosis and monitoring.		✓	
		I-02	Explain the relationship between chemical pathology and other pathology branches.			✓
		I-03	Describe the functions and working principles of biochemical apparatus used in chemical pathology.		✓	
		I-04	Illustrate the application of automated analyzers in patient diagnosis.			✓
		I-05	Explain the types, functions, and significance of reagents in clinical biochemistry.			✓
		I-06	Description, principle and operation of colorimeters - different types (Practical).			✓
J	Macronutrients and Micronutrients: Structure, Metabolism, and Functions	J-01	Define carbohydrates and describe their biological importance.		✓	
		J-02	Explain the digestion and major metabolic pathways of carbohydrates.			✓
		J-03	Identify the sources and types of lipids found in animal and plant-based foods.		✓	
		J-04	Describe lipid metabolism and explain its significance in energy			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			production.			
		J-05	List protein sources and identify their basic structural components.		✓	
		J-06	Explain protein digestion, metabolism, and the role of the urea cycle.			✓
		J-07	Define enzymes and describe their classification based on function.		✓	
		J-08	Explain enzyme action and the factors that influence enzyme activity.			✓
		J-09	Classify vitamins and identify their dietary sources.		✓	
		J-10	Discuss the role of vitamins in disease prevention and maintaining body function.			✓
K	Introduction to Blood Chemistry	K-01	Define blood chemistry and explain its significance in medical diagnostics.		✓	
		K-02	Identify key components typically measured in blood chemistry tests.		✓	
		K-03	Describe the importance of blood chemistry in assessing organ function and diagnosing diseases.			✓
		K-04	Explain the relationship between blood chemistry and metabolic processes.		✓	
		K-05	Interpret normal and abnormal			✓

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			glucose and HbA1c levels.			
		K-06	State normal ranges and clinical significance of BUN and creatinine.		✓	
		K-07	Discuss the significance of total protein and albumin in assessing nutritional and liver status.		✓	
		K-08	Compare HDL, LDL, total cholesterol, and triglycerides in lipid metabolism.		✓	
		K-09	Recognize normal ranges and functions of major blood electrolytes.		✓	
		K-10	Analyze the role of calcium in bone metabolism and its hormonal regulation.			✓
		K-11	Describe common liver enzymes and bilirubin as indicators of liver function.		✓	
		K-12	Correlate abnormal blood chemistry results with common clinical conditions.			✓
		K-13	Estimation of glucose in blood and interpretation of results (Practical).			✓
L	Electrolytes and Water	L-01	Identify key electrolytes in the human body (Na^+ , K^+ , Ca^{2+} , Cl^- , Fe) and their distribution.	✓		
		L-02	Describe the major physiological functions of each electrolyte.		✓	
		L-03	Explain the metabolic regulation		✓	

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			and hormonal control of electrolytes.			
		L-04	Correlate abnormal electrolyte levels with common clinical conditions.			✓
		L-05	Differentiate between intracellular and extracellular fluid compartments.		✓	
		L-06	Explain the physiological roles of water in the human body.		✓	
		L-07	Analyze the causes, types, and consequences of dehydration and oedema.			✓
M	Quantitative Analysis of Urine	M-01	Define quantitative urine analysis and its objectives.		✓	
		M-02	Identify the normal ranges of urea, creatinine, calcium, chloride, glucose, and albumin in urine.		✓	
		M-03	Describe the significance of measuring urea, creatinine, calcium, glucose, albumin, and chloride in urine.			✓
		M-04	Explain the procedure for collecting and measuring a 24-hour urine sample.			✓
		M-05	Calculate the total amount of a substance in urine using the standard formula.			✓
		M-06	List and recognize the instruments and reagents used in quantitative		✓	

	Content	SLO No.	SLOs Student Learning Outcomes	K	U	A
			urine analysis.			
		M-07	Discuss the clinical significance of quantitative urine tests in diagnosing and monitoring diseases.			✓

Ziauddin University Examination Board
Grade XII
Elementary Chemistry and Chemical Pathology
Table of Specification (TOS)

S.No	Domains	Weightage in assessment 100%	MCQs 1 mark each	PBAs 1 mark each	CRQs 4 marks each	ERQs 8 marks each
1	Introduction to Chemistry	1%	1	-	-	-
2	Elements	19%	1	-	2	2
3	Compounds and Mixtures	10%	1	-	1	1
4	Units of Measurements	4%	1	4	-	-
5	Solution	7%	1	4	1	-
6	Acid, Base and Salt	11%	2	-	1	1
7	Various Chemical Process	4%	1	-	1	-
8	Chemical Reactions	3%	-	-	1	-
9	Introduction to Chemical Pathology	6%	1	3	1	-
10	Macronutrients and Micronutrients: Structure, Metabolism, and Functions	12%	3	-	1	1
11	Introduction to Blood Chemistry	15%	3	4	1	1
12	Electrolytes and Water	4%	1	-	1	-
13	Quantitative Analysis of Urine	4%	1	-	1	-
Total # of Questions asked			17	15	12	6
Total # of Questions to be attempted			17	15	9	4
Maximum marks attainable			17 marks	15 marks	36 marks	32 marks

Ziauddin University Examination Board
Grade XII
Elementary Chemistry and Chemical Pathology
Scheme of Assessment

Maximum marks: 100

Section “A”

Multiple Choice Questions (MCQs)

(17 x 1 = 17)

- Attempt 17 MCQs: Each carries equal marks.

Practical based assessments (PBAs)

(15 x 1 = 15)

- Attempt 15 MCQs: Each carries equal marks.

Section “B” (Constructed Response Questions)

Constructed Response Questions (CRQs)

(9 x 4 = 36)

- Attempt any 9 questions out of 12. Each carries equal marks.

Section “C” (Extended Response Questions)

Extended Response Questions (ERQs)

(4 x 8 = 32)

- Attempt any 2 questions with its both sub parts (a and b) out of three questions.
- Each sub part consists of eight (08) marks.

DEFINITIONS OF COGNITIVE LEVELS

Remember

Remembering is the act of retrieving knowledge and can be used to produce things like definitions or lists. The student must be able to recall or recognise information and concepts. The teacher must present information about a subject to the student, ask questions that require the student to recall that information and provide written or verbal assessment that can be answered by remembering the information learnt.

Question Stems

- Can you name all the ...?
- Describe what happens when ...?
- How is (are) ...?
- How would you define ...?
- How would you identify ...?
- How would you outline ...?
- How would you recognise...?
- List the ... in order.
- What do you remember about ...?
- What does it mean?
- What happened after?
- What is (are) ...?
- What is the best one?
- What would you choose ...?
- When did ...?
- Where is (are) ...?
- Which one ...?
- Who spoke to ...?
- Who was ...?
- Why did ...?

Understand

The next level in the taxonomic structure is Understanding, which is defined as the construction of meaning and relationships. Here the student must understand the main idea of material heard, viewed, or read and interpret or summarise the ideas in their own words. The teacher must ask questions that the student can answer in their own words by identifying the main idea.

Question Stems

- Can you clarify...?
- Can you illustrate ...?
- Condense this paragraph.
- Contrast ...
- Does everyone think in the way that ... does?
- Elaborate on ...
- Explain why ...
- Give an example
- How can you describe...?
- How would you clarify the meaning...?
- How would you compare ...?
- How would you differentiate between.....?
- How would you describe...?
- How would you generalise...?
- How would you identify ...?
- Is it valid that ...?
- Is this the same as ...?
- Outline ...
- Select the best definition...
- State in your own words...
- This represents ...
- What are they saying?
- What can you infer from ...?
- What can you say about ...?
- What could have happened next?
- What did you observe?

	• What does this mean? • What expectations are there? • What information can you infer from...? • What is the main idea of ...? • What restrictions would you add? • What seems likely? • What seems to be ...? • What would happen if ...? • What might happen if ...? • Which are the facts? • Which statements support ...?
Apply The third level in Bloom's taxonomy, Applying, marks a fundamental shift from the pre-Bloom's learning era because it involves remembering what has been learnt, having a good understanding of the knowledge, and applying it to real-world exercises, challenges or situations. Students must apply an abstract idea in a concrete case to solve a problem or relate it to prior experience. The teacher must provide opportunities for students to use theories and problem-solving techniques in new situations and review and check their work. Assessment questions should be provided that allow students to define and solve problems. Question Stems • Can you group by characteristics such as ...? • Choose the best statements that apply... • Clarify why ... • Do you know of another instance where ...? • Draw a story map... • Explain why a character acted in the way that he did... • From the information given, can you develop a set of instructions about ...? • How would you develop ...? • How would you change ...? • How would you demonstrate...?	Analyse Analysing is the cognitive level where students can take the knowledge they have remembered, understood and applied, then delve into that knowledge to make associations, discernments or comparisons. Students should break down a concept or idea into parts and show relationships between these parts. Teachers must give students time to examine concepts and their requisite elements. Students are required to explain why they chose a solution. Question Stems • Can you distinguish between ...? • Can you explain what must have happened when ...? • Determine the point of view, bias, values, or intent underlying the presented material... • Discuss the pros and cons of ... • How can you classify ... according to ...? • How can you compare the different parts? • How can you sort the different parts...? • How is ... connected to ...? • How is ... similar to ...? • How would you categorise...? • How would you explain...?

• How would you develop? • How would you explain ...? • How would you modify ...? • How would you present...? • How would you solve ...? • Identify the results of ... • Illustrate the ... • Judge the effects of ... What would result ...? • Predict what would happen if ... • Tell how much change there would be if ... • Tell what would happen if ... • What actions would you take to perform ...? • What do you think could have happened next? • What examples can you find that? • What other way would you choose to ...? • What questions would you ask of ...? • What was the main idea ...? • What would the result be if ...? • Which factors would you change if ...? • Who do you think...? • Why does this work? • Write a brief outline ... • Write in your own words ...	• What could the ending have been if ... had taken place? • State the point of view of ... • What are some of the problems of ...? • What assumptions ...? • What can you infer about...? • What can you point out about....? • What conclusions ...? • What do you see as other possible outcomes? • What does the author assume? • What explanation do you have for ...? • What ideas justify the conclusion? • What ideas validate...? • What is the analysis of ...? • What is the function of ...? • What is the problem with ...? • What motive is there? • What persuasive technique is used? • What statement is relevant? • What was the turning point? • What were some of the motives behind ...? • What's fact? Opinion? • What's the main idea? • What's the relationship between? • Which events could not have happened? • Why did ... changes occur? • Why do you think ?
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BLOOM'S TAXONOMY WITH EXAMPLES

If you are a teacher looking for ways to engage your students in learning, this LIST of questions might be interesting for your classroom practice. Bloom's Taxonomy question stems can help elicit higher-order thinking skills and promote critical thinking among learners at different taxonomy levels. These question stems can also encourage students to think about their knowledge through reflection before answering questions.

ACTION WORDS FOR COGNITIVE LEVELS

Knowledge	Understand	Apply	Analyze	Evaluate	Create
					
define	explain	solve	analyze	reframe	design
identify	describe	apply	appraise	criticize	compose
describe	interpret	illustrate	judge	evaluate	create
label	paraphrase	modify	support	order	plan
list	summarize	use	compare	compare	combine
name	classify	calculate	decide	classify	formulate
state	compare	change	discriminate	contrast	invent
match	differentiate	choose	recommend	distinguish	hypothesize
recognize	discuss	demonstrate	summarize	infer	substitute
select	distinguish	discover	assess	separate	write
examine	extend	experiment	choose	explain	compile
locate	predict	relate	convince	select	construct
memorize	associate	show	defend	categorize	develop
quote	contrast	sketch	estimate	connect	generalize
recall	convert	complete	grade	differentiate	integrate
reproduce	demonstrate	construct	measure	divide	modify
tabulate	estimate	dramatize	predict	order	organize
tell	express	interpret	rank	prioritize	prepare
Copy	identify	manipulate	score	survey	produce

discover	indicate	paint	select	calculate	rearrange
duplicate	infer	prepare	test	conclude	rewrite
enumerate	relate	teach	argue	correlate	adapt
listen	restate	act	conclude	deduce	anticipate
observe	select	collect	consider	devise	arrange
omit	translate	compute	critique	diagram	assemble
read	ask	explain	debate	dissect	choose
recite	cite	list	distinguish	estimate	collaborate
record	discover	operate	editorialize	evaluate	facilitate
repeat	generalize	practice	justify	experiment	imagine
retell	group	simulate	persuade	focus	intervene
visualize	illustrate	transfer	rate	illustrate	make
	judge	write	weigh	organize	manage
	observe			outline	originate
	order			plan	propose
	report			question	simulate
	represent			test	solve
	research				support
	review				test
	rewrite				validate
	show				

HSSC PART II EXAMINATION MARKS BREAKUP GRID FOR EXAMINATION 2026

GROUP: PRE-MEDICAL

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
PHYSICS	85	15	100
CHEMISTRY	85	15	100
BIOLOGY	85	15	100
TOTAL	505	45	550

GROUP: PRE-ENGINEERING

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
PHYSICS	85	15	100
CHEMISTRY	85	15	100
MATHEMATICS	100	--	100
TOTAL	520	30	550

GROUP: GENERAL SCIENCE

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
PHYSICS	85	15	100
COMPUTER SCIENCE	75	25	100
MATHEMATICS	100	--	100
TOTAL	510	40	550

GROUP: COMMERCE

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
ECONOMICS	75	-	75
P.O.C	75	-	75
ACCOUNTING	100	--	100
BUSINESS MATHEMATICS	50		50
TOTAL	550	---	550

GROUP: HUMANITIES

(Any Three Electives)

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
COMPUTER SCIENCE	75	25	100
ISLAMIC STUDIES	100		100
MATHEMATICS	100	-	100
SOCIOLOGY	100	--	100
ECONOMICS	100		100
EDUCATION	100		100
CIVICS	100		100
NURSING	85	15	100
TOTAL	550	---	550

GROUP: MEDICAL TECHNOLOGY

SUBJECT	THEORY	PBA	TOTAL
ENGLISH	100	-	100
URDU NORMAL / SINDHI NORMAL	100	-	100
PAKISTAN STUDIES	50	-	50
MICROBIOLOGY	85	15	100
CLINICAL PATHOLOGY & SEROLOGY	85	15	100
ELEMENTARY CHEMISTRY & CHEMICAL PATHOLOGY	85	15	100
TOTAL	505	45	550